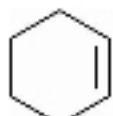
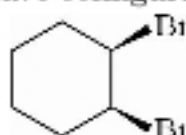
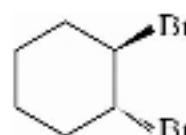
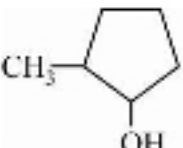
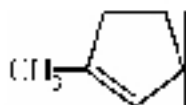
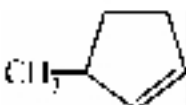
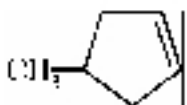
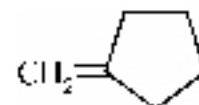


- *38. The reagent/s required for the conversion $R_1CH = CHR_2 \longrightarrow R_1COOH + HOOCR_2$ is (are) :
- (A) O_3 followed by treatment with H_2O_2
 (B) Baeyer's reagent
 (C) Hot $KMnO_4/KOH$ followed by acidification
 (D) Both (A) and (C)
39. An organic compound with molecular formula C_6H_{12} upon ozonolysis gives only acetone as the product. The compound is :
- (A) 2, 3-dimethyl-1-butene (B) 3-hexene
 (C) 2-hexene (D) 2, 3-dimethyl-2-butene
40. An alkene having the molecular formula C_9H_{18} on ozonolysis gives 2, 2-dimethylpropanal and 2-butanone. The alkene is :
- (A) 2, 2, 2-trimethyl-3-hexene (B) 2, 2, 6-trimethyl-3-hexene
 (C) 2, 3, 4-trimethyl-2-hexene (D) 2, 2, 4-trimethyl-3-hexene
41.  + $Br_2 \longrightarrow A$
 A will have configuration :
- (A)  (B) 
 (C) Both (A) and (B) (D) None of the above
42.  $\xrightarrow[\Delta]{H^+}$ A (major) , A is :
- (A)  (B)  (C)  (D) 
43. Aqueous solution of an organic compound, 'A' on electrolysis liberates acetylene and CO_2 at anode. 'A' is:
- (A) potassium acetate (B) potassium succinate
 (C) potassium citrate (D) potassium maleate
44. Acetylene can be obtained by the reaction :
- (A) $HCOOK \xrightarrow{\text{Electrolysis}}$ (B) $CHI_3 + Ag \xrightarrow{\Delta}$
 (C) $CH_3CH_2OH \xrightarrow[443\text{ K}]{\text{Conc. } H_2SO_4}$ (D) $Be_2C + H_2O \longrightarrow$
45. In the series, $C_2H_2 \xrightarrow[2.0\text{ eq.}]{NaNH_2} X \xrightarrow[2.0\text{ eq.}]{CH_3I} Y \xrightarrow[H_2SO_4, H_2O]{HgSO_4} Z$ the compound Z is :
- (A) $CH_3CH_2CH = CH_2$ (B) CH_3COCH_3
 (C) $CH_3 - \overset{\overset{O}{\parallel}}{C} - CH_2CH_3$ (D) $CH_3CH_2CH_2CHO$